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CHEMISTRY SEM-3 MDL-3Polymers

polymers are large macromolecules formed by joining many small repeating units called monomers.

Polymers formed by the polymerization of monomers.

classification of polymers.

- (1) Natural polymers:- It is obtained from nature. Such as plant and animal.
eg. Starch, cellulose, proteins and natural rubber etc.
- (2) Synthetic polymers:- This type of polymers made in laboratory.
eg. polythene, nylon, synthetic rubber etc.
- (3) Semi-synthetic:- It is derived from natural polymers.
eg. cellulose acetate, rayon etc.
- (4) Addition polymers:- It is formed by self addition of monomers without losing any small molecules.
eg. Polyethylene, PVC, Teflon etc.

(5) condensation polymers: ~~it is formed by the combination of monomers with the elimination of simple small molecules like water or methanol.~~ It is formed by the combination of monomers with the elimination of simple small molecules like water or methanol.

eg. Nylon 6,6, Bakelite

(6) Elastomers: - weak intermolecular forces; can be stretched.

eg. Buna-S, Buna-N, ^{Natural} rubber.

(7) Fibres: - strong intermolecular forces like hydrogen bonding, high tensile strength.

eg. Nylon 6,6, Terylene.

(8) Thermoplastics: - intermolecular force between elastomers and fibres; can be repeatedly melted and reshaped - eg. Polythene, PVC.

(9) Thermosetting plastics - Heavily branched or cross linked network polymers become hard on heating and cannot be reused.
eg. Bakelite

